

UNIT – 3

Acidifiers:

1. Ammonium Chloride (NH₄Cl):

- **Preparation:** Prepared by the reaction between ammonia (NH₃) and hydrochloric acid (HCl).
- **Assay:** Assayed by titration with a strong base, such as sodium hydroxide (NaOH).
- **Properties:** It is a white crystalline powder with a salty taste.
- **Medicinal Uses:**
 - Used to treat metabolic alkalosis, a condition characterized by high blood pH.
 - Acts as an expectorant in cough preparations to help clear mucus from the airways.

2. Dilute Hydrochloric Acid (Dil. HCl):

- **Preparation:** Prepared by diluting concentrated hydrochloric acid with water.
- **Assay:** Assayed by titration with a standard alkali solution (sodium hydroxide, NaOH) or assessed based on its specific gravity.
- **Properties:** It is a colourless, odourless liquid.
- **Medicinal Uses:**
 - Used as a gastric acidifier to aid in the digestion process.
 - An adjunct in the treatment of certain gastrointestinal disorders to lower the pH in the stomach.

Antacids:

• Ideal Properties of Antacids:

- Rapid onset of action to provide quick relief.
- Prolonged duration of action for sustained relief.
- High neutralizing capacity.
- Minimal adverse effects or interactions.
- Good patient acceptability (e.g., pleasant taste).

• Common Antacids:

• Sodium Bicarbonate (NaHCO₃):

- **Properties:** It is a white, crystalline powder with a slightly salty taste.

- Acts as a fast-acting antacid but may lead to metabolic alkalosis if overused.
- **Aluminium Hydroxide Gel:**
 - Provides a slow onset but long-lasting antacid effect.
- **Magnesium Hydroxide Mixture:**
 - Acts moderately quickly and has a moderate duration of action.
 - Often used in combination with aluminium hydroxide for balanced efficacy.

Cathartics:

- **Common Cathartics:**
 - **Magnesium Sulphate:**
 - A saline cathartic that works by drawing water into the intestines and stimulating bowel movements.
 - **Sodium Orthophosphate:**
 - Acts as an osmotic cathartic, increasing the water content in the intestines.
 - **Kaolin and Bentonite:**
 - These are absorbent substances used as mild laxatives, particularly for treating diarrhoea.

Antimicrobials:

- **Mechanism of Antimicrobials:**
 - Antimicrobials inhibit the growth or kill microorganisms such as bacteria, fungi, and viruses.
 - They may work through mechanisms like disrupting cell walls, interfering with DNA replication, or inhibiting enzyme function.
- **Classification:**
 - Antimicrobials can be classified into various categories, including antibiotics, antiseptics, and disinfectants, depending on their specific mechanisms and applications.
- **Common Antimicrobial Agents:**
 - **Potassium Permanganate:**
 - Used as an antiseptic and disinfectant to clean wounds and surfaces.
 - **Boric Acid:**

- Exhibits antifungal and antiseptic properties, often used in ointments and eye washes.
- **Hydrogen Peroxide (H₂O₂):**
 - Used for wound cleaning and as a disinfectant to prevent infection.
- **Chlorinated Lime (Calcium Hypochlorite):**
 - Used as a disinfectant for water purification and surface sanitation.
- **Iodine and Its Preparations:**
 - Iodine tincture and iodophors (iodine-containing compounds) are used as antiseptics and disinfectants.

PHARMACY PEERS